

## RECENT TRENDS IN DEVELOPMENT OF HIGH VOLTAGE CIRCUIT BREAKERS WITH SF<sub>6</sub> ALTERNATIVE GASES

M. SEEGER<sup>a,\*</sup>, R. SMEETS<sup>b</sup>, J. YAN<sup>c</sup>, H. ITO<sup>d</sup>, M. CLAESSENS<sup>a</sup>, E. DULLNI<sup>e</sup>,  
L. FALKINGHAM<sup>f</sup>, C. M. FRANCK<sup>g</sup>, F. GENTILS<sup>h</sup>, W. HARTMANN<sup>i</sup>,  
Y. KIEFFEL<sup>j</sup>, S. JIA<sup>k</sup>, G. JONES<sup>c</sup>, J. MANTILLA<sup>a</sup>, S. PAWAR<sup>l</sup>, M. RABIE<sup>g</sup>,  
P. ROBIN-JOUAN<sup>j</sup>, H. SCHELLEKENS<sup>h</sup>, J. SPENCER<sup>c</sup>, T. UCHII<sup>m</sup>, X. LI<sup>k</sup>,  
S. YANABU<sup>k</sup>

<sup>a</sup> ABB Switzerland Ltd, Corporate Research Center, Segelhofstrasse 1K, CH-5405 Baden-Dättwil, Switzerland

<sup>b</sup> DNV GL, Utrechtseweg 310, 6800 ET, Arnhem, Netherlands

<sup>c</sup> University of Liverpool, Department of Electrical Engineering and Electronics, Liverpool, L69 3GJ, UK

<sup>d</sup> Mitsubishi Electric Corp., Tokyo Building 2-7-3, Marunouchi, Chiyoda-ku, Tokyo 100-8310, Japan

<sup>e</sup> ABB AG, Bahnstrasse 39-4, D-40832 Ratingen, Germany

<sup>f</sup> Vacuum Interrupters Limited, Sir Frank Whittle Business Centre, Great Central Way, Rugby CV21 3XH, UK

<sup>g</sup> ETH Zürich, Rämistrasse 101, 8092 Zürich, Switzerland

<sup>h</sup> Schneider Electric, 35, rue Joseph Monier, 92500 Rueil-Malmaison, France

<sup>i</sup> Siemens AG, Corp. Techn., Otto-Hahn-Ring 6, 81739 Munich, Germany

<sup>j</sup> GE Grid Solutions, 51 Espl. du Général de Gaulle, 92900 Puteaux, France

<sup>k</sup> Xi'an Jiaotong University, 28 Xianning West Road, Beilin, Xi'an, Shaanxi, China

<sup>l</sup> Crompton Greaves, Dr. Annie Besant Road, Worli, Mumbai – 40001, India

<sup>m</sup> Toshiba Corp., Power and Industrial Systems R&D Center, 8 Shinsugita-cho, Kanagawa 235-8523, Japan

\* martin.seeger@ch.abb.com

**Abstract.** The available knowledge of state-of-the-art of SF<sub>6</sub> alternative gases in switching applications was collected and evaluated in an initiative of the Current Zero Club [1] together with CIGRE. The present contribution summarizes the main results of this activity and will also include the latest trends. The main properties and switching performance of new gases are compared to SF<sub>6</sub>. The most promising new gases are at the moment perfluoroketones and perfluoronitriles. Due to the high boiling point of these gases, in HV applications mixtures with CO<sub>2</sub> are used. For MV insulation perfluoroketones are mixed with air, but also other combinations might be possible. The dielectric and switching performance of the mixtures, with mixing ratios that allow sufficiently low operating temperatures, is reported to be only slightly below SF<sub>6</sub>. Minor design changes or de-rating of switchgear are therefore necessary. Differences between the gas mixtures are mainly in the boiling point and the GWP.

**Keywords:** SF<sub>6</sub> alternative gases, CO<sub>2</sub>, Circuit Breaker.

## 1. Introduction

SF<sub>6</sub> is widely used in electric power transmission and distribution systems, as for example in gas insulated switchgear (GIS), circuit breakers (CB) and medium voltage (MV) load break switches. It combines unique electrical insulation and arc interruption capability [2]. However, it is also a very strong greenhouse gas with a global warming potential (GWP) of about 23500 over a time horizon of 100 years, e.g. [3] and its use is regulated and restrictions are discussed. Therefore, search for alternative gases for use in power applications has been ongoing since about two decades ago e.g. [4, 5]. The state of the art of SF<sub>6</sub> alternative gases for switching applications was recently addressed in an initiative of the current zero club (CZC) [1] in collaboration with CIGRE. A survey was done collecting all the available recent literature on the topic. The result

was presented and discussed at a joint workshop at the CIGRE session 2016. The present paper gives the main results of this survey. Since vacuum switching technology is a separate ongoing activity [6], it will be left out in the present review.

## 2. Alternative gases

The intensification of search for alternative gases started about two decades ago [4, 5] after the Kyoto protocol was agreed in 1997 and further increased in the last 10 years (e.g. [7–15]). Important requirements for alternative gases were identified as: Low GWP, zero ozone depletion (ODP) potential, low toxicity, non-flammability, high dielectric strength, high arc quenching and heat dissipation capability, stability and material compatibility and availability on market.

From various studies of gases of natural origin, CO<sub>2</sub>

	CAS number	Boiling point °C	GWP	ODP	Flamm- ability	Toxicity LC50 (4h) ppmv	Toxicity TWA <sup>1)</sup> ppmv	Dielectric strength at 0.1 MPa	Ref
SF <sub>6</sub>	2551-62-4	-64 <sup>2)</sup>	23500	0	No	-	1000	1	[7, 16]
CO <sub>2</sub>	124-38-9	-78.5 <sup>2)</sup>	1	0	No	>300000	5000	≈0.3	[4, 5, 17]
C5-PFK	756-12-7	26.5	<1	0	No	≈20000	225	≈2	[13, 16, 18]
C4-PFN	42532-60-5	-4.7	2100	0	No	12000...15000	65	≈2	[7, 16, 19-21]

<sup>1)</sup> The occupational exposure limit is given by a time-weighted-average (TWA), 8-hr

<sup>2)</sup> Sublimation point

Table 1. Properties of pure gases compared to SF<sub>6</sub>.

turned out to be the most promising arc quenching gas, e.g. [8, 11], possibly enhanced in performance by some additives [12], like e.g. O<sub>2</sub> or CF<sub>4</sub>. However, as was shown, the switching and dielectric performances of CO<sub>2</sub> are both below those of SF<sub>6</sub>, e.g. [11, 17]. Other interesting gases were identified to be fluorinated gases like CF<sub>3</sub>I, hydrofluoroolefins (HFO1234ze and HFO1234yf), perfluoroethanes (e.g. C5F10O), perfluoronitriles (C<sub>4</sub>F<sub>7</sub>N), fluoroethers (HFE245cb2), fluorooxiranes and hydrochlorofluoroolefins (HCFO1233zd), e.g. [7, 13-16, 22]. Taking all the requirements into account, the most promising candidates at present appeared to be the C5 perfluoroketone (CF<sub>3</sub>C(O)CF(CF<sub>3</sub>)<sub>2</sub> or C5-PFK) [18] and the iso-C4 perfluoronitrile ((CF<sub>3</sub>)<sub>2</sub>-CF-CN or C4-PFN) [19]. The dielectric performance of pure gases scales with the boiling point, i.e. gases with high dielectric strength usually also have a high boiling point, see e.g. [10]. For C5-PFK and C4-PFN, the boiling points at 0.1 MPa are 26.5 °C and -4.7 °C, respectively. Thus, for application in switchgear, where a sufficiently low boiling point is needed for low temperature requirements, an admixture of a buffer gas is needed. CO<sub>2</sub> is selected for this role in HV due to its good arc quenching capability, e.g. [9, 11, 12]. In MV application air is also reported as the buffer gas in combination with C5-PFK for insulation purposes [23-25]. The concentration of C5-PFK and C4-PFN, and by this the performance of the mixtures, will depend on the minimum operating temperature requirement of the switchgear. An additional alternative approach is proposed to use air for insulation and vacuum CB (VCB) for switching [26, 27].

### 3. Properties of pure gases and mixtures

The properties of the selected alternative gases with reference to SF<sub>6</sub> are shown in table 1. The GWP for the various gases are different: the C4-PFN has a much higher GWP than CO<sub>2</sub> or C5-PFK that are both around 1. All the gases of interest are not flammable, have no ODP and are non-toxic according to technical and safety data sheets available from the chemical manufacturer [18-20, 28, 29]. The dielectric strength of pure C4-PFN and C5-PFK is nearly twice that of SF<sub>6</sub>. CO<sub>2</sub> has a dielectric withstand comparable to air [4, 17], i.e. significantly below that of SF<sub>6</sub>. The prop-

erties of gases and mixtures when used in switchgear are shown in table 2. The concentration of admixtures of C4-PFN and C5-PFK with the buffer gas is given in the second column and is typically below 13 % (molar concentration). Note that for the use of C5-PFK in CO<sub>2</sub> additionally an oxygen admixture is reported, since the presence of oxygen reduces the generation of harmful by-products like CO and solid by-products such as soot [30]. Due to a reduced dielectric withstand of the mixtures compared to SF<sub>6</sub> (column 6) at the same pressure the minimum operating pressure needs to be increased to about 0.7...0.8 MPa for C5-PFK and C4-PFN when using CO<sub>2</sub> as the buffer gas for HV application, see column 3 in table 2. For Air/C5-PFK mixtures in MV application 0.13 MPa can be kept and the dielectric withstand of SF<sub>6</sub> is approached. The high dielectric withstand of mixtures with relatively low admixture ratios of C4-PFN or C5-PFK can be explained by a synergy effect [7, 30, 31], i.e. a non-linear increase of the dielectric strength with the admixture ratio, as it is known for SF<sub>6</sub>/N<sub>2</sub> mixtures [32]. The GWP of mixtures with C5-PFK is negligible, at the cost of a higher minimum operating temperature. Low temperature applications of e.g. -25 °C for HV can be covered by pure CO<sub>2</sub> or CO<sub>2</sub>+C4-PFN mixtures. This is at the cost of significantly reduced dielectric withstand in case of pure CO<sub>2</sub> or significantly higher GWP in case of C4-PFN mixtures. Due to strong dilution, the toxicity of the mixtures is well below that of the pure substances, see e.g. [7, 33].

### 4. Switching performance of alternative gases

Preliminary information on the switching performance of pure CO<sub>2</sub> and CO<sub>2</sub> mixtures is collected in table 3. The performance of SF<sub>6</sub> is given for comparison. With an enhanced operating pressure compared to SF<sub>6</sub> the cold dielectric strength, which is e.g. a measure of the performance in capacitive switching, can reach that of SF<sub>6</sub>.

In the scanned literature, only qualitative statements on the switching performance of C4-PFN and C5-PFK mixtures could be found. For CO<sub>2</sub> a few quantitative comparisons exist. Very roughly, for pure CO<sub>2</sub> at an increased filling pressure of about 1 MPa, about 2/3 of the dielectric and thermal interruption

	$C_{ad}$ <sup>1)</sup>	$p_{min}$ <sup>2)</sup>	$T_{min}$ <sup>3)</sup>	GWP	D.S. <sup>4)</sup>	Toxicity LC50 ppmv	Ref
		MPa	°C				
SF <sub>6</sub>	-	0.43...0.6	-41...-31	23500	0.86...1	-	
CO <sub>2</sub>	-	0.6...1	≤ -48 <sup>6)</sup>	1	0.4...0.7	> 3e5	[8, 11, 12]
CO <sub>2</sub> /C5-PFK/O <sub>2</sub> (HV)	≈ 6/12	0.7	-5...+5	1	≈ 0.86	> 2e5	[13, 22, 27, 30]
CO <sub>2</sub> /C4-PFN (HV)	≈ 4...6	0.67...0.88	-25...-10	327...690	0.87...0.96	> 1e5	[7, 15, 21, 33, 34]
Air/C5-PFK (MV)	≈ 7...13	0.13	-25...-15	0.6	≈ 0.85 <sup>5)</sup>	1e5	[16, 24, 25]
N <sub>2</sub> /C4-PFN (MV)	≈ 20...40	0.13	-25...-20	1300...1800	0.9...1.2	> 2.5e4	[15]

<sup>1)</sup> Concentration of admixture is in mole % referred to the gas mixture

<sup>2)</sup> Typical lock out pressure range

<sup>3)</sup> Minimum operating temperature for  $p_{min}$

<sup>4)</sup> Dielectric strength compared to SF<sub>6</sub> at 0.55 MPa. For the scaling of SF<sub>6</sub> breakdown field  $E_d$  with pressure correction in the form of  $E_d = 84p^{0.71}$  was used [32]

<sup>5)</sup> Compared to SF<sub>6</sub> at 0.13 MPa, measurements were for a mixture at -15 °C

<sup>6)</sup> Calculations with Refprop: <https://www.nist.gov/srd/refprop>

Table 2. Properties/performances of pure gases and mixtures in MV and HV switchgear applications.

performance of SF<sub>6</sub> might be expected. With the admixture of O<sub>2</sub> to CO<sub>2</sub> in the mixing ratio range up to 30 %, an increase of the thermal interruption performance [12] and also a slight increase in dielectric strength (e.g. [35]) is expected. With the admixture of C4-PFN and C5-PFK into CO<sub>2</sub> the dielectric performance can be close to SF<sub>6</sub>. The short-line fault (SLF) switching performance for the mixtures of CO<sub>2</sub>/O<sub>2</sub>/C5-PFK is reported to be 20 % below that of SF<sub>6</sub> [30]. For an adapted CB with CO<sub>2</sub>/C4-PFN a similar SLF performance to that of SF<sub>6</sub> is stated, e.g. [7]. There are, however, also direct comparisons of pure CO<sub>2</sub> with CO<sub>2</sub>/C4-PFN and CO<sub>2</sub>/C5-PFK mixtures using identical geometry and pressure, which show similar thermal interruption performance of CO<sub>2</sub> with and without admixtures [27]. IEC test duties L90 (SLF) and T100 (100 % terminal fault) with the new mixtures are passed with some design modifications [36] or certain de-rating [30], suggesting that the switching performance of the new mixtures is not significantly lower than that of SF<sub>6</sub>. This has also been shown to be valid for the bus transfer switching duty of disconnecter switches, e.g. [36, 37]. It is expected that dedicated design improvements can still increase the switching performance in the future.

An important point is the toxicity of the gas after arcing. C5-PFK and C4-PFN are complex molecules which start to decompose above approximately 650 °C in case of C4-PFN, e.g. [32]. After decomposition C5-PFK and C4-PFN molecules do not recombine to their original structure, but form smaller molecules. A decomposition rate of 0.5 Moles/MJ under high current switching is reported for CO<sub>2</sub>/O<sub>2</sub>/C5-PFK mixtures [30]. For partial discharges decomposition rates of more than one order of magnitude lower are observed for this mixture [38]. No quantitative information is given so far on the decomposition rates of C4-PFN. Note that this decomposition involving the new gases is not comparable with the decomposition of SF<sub>6</sub> because the latter only occurs due to chemical reactions with ablated contact and nozzle material. The decomposition involving the new gases is not seen as a problem over lifetime, but concentrations in the

equipment need to be monitored or regularly checked, in a way similar to SF<sub>6</sub> [39]. Most toxic decomposition products for HV, i.e. mixtures with CO<sub>2</sub>, are CO and HF, e.g. [30, 32]. The arced mixtures are regarded to have similar or lower toxicity as arced SF<sub>6</sub>. It is recommended, therefore, to treat this in a way similar to arced SF<sub>6</sub>. It must, however, be noted that the above statement is made only based on the limited knowledge available on the toxicity of the new gases. Formation of critical by-products under repetitive switching in a small volume is discussed in [16]. Considerable more experience seems to be needed on the post arcing toxicity of the potential SF<sub>6</sub> substitute gases. Additional reported issues are: material compatibility [22, 32] (e.g. effects on sealings and grease), gas tightness and gas handling procedures. Therefore, it should not be expected that existing HV equipment can be filled with the new gases without design or material changes. Internal arc tests were done with all mixtures and no critical issues are reported, e.g. [7, 22, 24]. Heat dissipation of the mixtures is slightly inferior to SF<sub>6</sub> [7, 22], i.e. moderate de-rating or design changes might be necessary with respect to the current carrying capability. At present, field experience is gained with CO<sub>2</sub> live-tank CB [40], being started some years ago. A CO<sub>2</sub> filled CB is also commercially available [41]. With the C5-PFK mixtures for HV and MV pilot installations have been in operation successfully since 2015 in Switzerland [22, 39] and Germany [42]. Pilot installations with the CO<sub>2</sub>/C4-PFN mixture are planned in several European countries [7], such as a 145 kV indoor GIS in Switzerland, 245 kV outdoor Current Transformers in Germany and outdoor 420 GIL in UK and Scotland [7, 34, 36].

## 5. Conclusions and outlook

Published information on alternative gases for SF<sub>6</sub> in switching applications has been reviewed. In their present state, these investigations have just started and are by far not as extensive as for SF<sub>6</sub>. The presently available manufacturer information on prop-

	Operating pressure MPa	Dielectric strength	SLF performance compared to SF <sub>6</sub> <sup>1)</sup>	Dielectric reco- very speed	Ref
SF <sub>6</sub>	0.6	1	1	1	
CO <sub>2</sub>	0.8...1	0.5...0.7	0.5...0.83	≥ 0.5	[8, 11, 12, 27]
CO <sub>2</sub> +C5-PFK/O <sub>2</sub>	0.7...0.8	close to SF <sub>6</sub>	0.8...0.87	close to SF <sub>6</sub>	[22, 27]
CO <sub>2</sub> /C4-PFN	0.67...0.82	close to SF <sub>6</sub>	0.83... (1) <sup>2)</sup>	close to SF <sub>6</sub>	[7, 27, 32]

<sup>1)</sup> At same pressure build up

<sup>2)</sup> Same performance as SF<sub>6</sub> is stated but it is not clear if this was under same condition

Table 3. Switching performance of gases and mixtures compared to SF<sub>6</sub> at increased operating pressures in HV applications

erties shows that new gases (e.g. C5-PFK and C4-PFN) are available, which can compete with, but may not fully reach the performance of, SF<sub>6</sub> when used in mixture with CO<sub>2</sub> as the buffer gas. Main differences are in the insulation and interruption performances and boiling point with the latter defining the minimum operating temperature specified for the switchgear. The lowest operating temperatures (e.g. −50 °C) can be reached with CO<sub>2</sub>. However, CO<sub>2</sub> seems to have an overall lower interruption performance, especially in dielectric interruption and withstand, than gas mixtures containing C4-PFN or C5-PFK. The advantage of CO<sub>2</sub>/C5-PFK mixture compared with CO<sub>2</sub>/C4-PFN mixture is the negligible GWP of about 1 compared to 427...600 of the latter. The advantage of CO<sub>2</sub>/C4-PFN compared to CO<sub>2</sub>/C5-PFK is the lower minimum operating temperature of about −25 °C compared to about −5 °C of the latter. Since research and development of these new SF<sub>6</sub> alternatives has just started, design improvements can be expected in the future. Exhaustive studies on decomposition products after current switching and their level of toxicity are still required, as it was performed in the past for SF<sub>6</sub>, in different operating conditions. Probably from all different alternatives, a convergence to a single solution can be expected on the longer term. For sure, much more investigations and experimental validations have to be carried out.

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